



The near-IR absorption spectrum of HD¹⁸O, HD¹⁶O, D₂¹⁸O and D₂¹⁶O using Fourier-transform incoherent broadband cavity enhanced absorption spectroscopy

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Using a combination of Fourier-transform (FT) spectroscopy and incoherent broad-band cavity-enhanced absorption spectroscopy (IBBCEAS) [1] very small optical densities can be measured with high spectral resolution over a wide spectral region (several hundred cm^{-1}) in a small sample volume. In the near infra-red these experimental features enable the detection of atmospheric trace constituents with high selectivity [2]. We applied FT-IBBCEAS to a sample mixture of 8.0 mbar of D₂¹⁶O and 12.4 mbar of pure H₂¹⁸O in the optical cavity. The spectral range covered was 5800 cm^{-1} to 7000 cm^{-1} at a resolution of 0.02 cm^{-1} . Even though the water isotope H₂¹⁸O has been studied before in this spectral region by different techniques with typical absorption path lengths of up to \sim 100 m [3], the superior absorption sensitivity of FT-IBBCEAS in combination with D₂O being present in the cavity results in the observation of a large number of new absorption features in the spectrum which are notably due to the rare HD¹⁸O isotope of water. To analyse the spectrum line-lists for each of the six possible water isotopologues have been established using a uniform intensity threshold which gave a pool of 5547 candidate transitions. 3576 lines in the spectrum have assignments and 1971 lines are unaccounted for. The following spectral features were identified. H₂¹⁸O: 346 assigned from 704 above threshold, H₂¹⁶O: 304 from 566; D₂¹⁶O: 233 from 295 (some of which are new); D₂¹⁸O: 254 from 338 (all of these are new assignments). HD¹⁶O: 1247+ from 1692; HD¹⁸O: 1192+ from 1952, with \sim 985 fully labelled, only 415 of these are listed in IUPAC (i.e. the upper state energy levels were already known) [4,5] (+: about 250 lines are blurred between HD¹⁶O and HD¹⁸O).

The measurements demonstrate the usefulness of this experimental approach for spectroscopic investigations of isotopic or very dangerous samples, electric or plasma discharges or flames.

[1] A.A. Ruth, J. Orphal, S. E. Fiedler, "Cavity Enhanced Fourier Transform Absorption Spectroscopy using an Incoherent Broadband Light Source", *Appl. Opt.*, **46** (2007) 3611-3616.

[2] J. Orphal, A. A. Ruth, "High-resolution Fourier-transform cavity-enhanced absorption spectroscopy in the near-infrared using an incoherent broad-band light source", *Opt. Express*, **16** (2008) 19232-19243.

[3] A.-W. Liu, O. Naumenko, K.-F. Song, B. Voronin, S.-M. Hu, "Fourier-transform absorption spectroscopy of H₂¹⁸O in the first hexade region", *J. Mol. Spectrosc.* **236** (2006) 127-133.

[4] IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part I- Energy levels and transition wavenumbers for H₂¹⁷O and H₂¹⁸O, J. Tennyson et al., *JQSRT*, **110** (2009) 573-596.

[5] IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part I- Energy levels and transition wavenumbers for HD¹⁶O, HD¹⁷O and HD¹⁸O, J. Tennyson et al., *JQSRT*, **111** (2010) 2160-2184.